Tetsuji Kametani,* Yasuyuki Kato, Fumio Satoh, and Keiichiro Fukumoto: Studies on the Syntheses of Heterocyclic Compounds. 696. Stereochemistry of Four Isomeric 4a-Cyano-1,2,3,4,4a,9,10,10a-octahydro-7-methoxy-1-(methoxycarbonyl)-1methylphenanthrenes.

Page 1178. Column 1. Structures B and D in Scheme II should be as follows:

$$R^1$$
 OMe

Page 1178. Column 2, line 20 from bottom. "Wenkert 12 had proposed that podocarpic acid . . ." should read "On the basis of the experiment using podocarpic acid, Wenkert12 had proposed that

F. Ramirez,* E. Evangelidou-Tsolis, A. Jankowski, and J. F. Marecek: Synthesis of Deoxyribooligonucleotides by Means of Cyclic Enediol Pyrophosphates.

Page 3145, Scheme I. Formula 8a should read:

Thy
$$HO \longrightarrow P \longrightarrow O$$

$$O^{-}M^{+}$$

$$8a M^{+} = (C.H.) NH^{+}$$

W. H. Pirkle* and P. L. Rinaldi: Nuclear Magnetic Resonance Determination of Enantiomeric Compositions of Oxaziridines Using Chiral Solvating Agents.

Page 3217. The sense of nonequivalence listed for R2 of compound la is incorrectly listed in Table I as H (high field). It should read L

Antonio Focella,* Sidney Teitel, and Arnold Brossi: A Simple and Practical Synthesis of Olivetol.

Page 3457. Column 1, paragraph 2. The physical data values listed under Olivetol (3) should be deleted and replaced with the following: "An analytical sample crystallized from water gave: mp 39–41 °C (lit. 8,21 mp 39–41 °C); NMR (Me₂SO) δ 0.85 (CH₃), 1.28 (2CH₂), 1.49

(CH₂), 2.36 (CH₂), 3.44 (H₂O), 6.53 (aromatics), 8.92 (OH)."

David T. Connor,* Clive Greenough, and Maximillian von Strandtmann: W-7783, a Unique Antifungal Antibiotic.

Page 3665. Scheme I. There is an error in 1 and 2. The relative stereochemistry for ring C should be as follows:

X. Creary* and A. J. Rollin: Rearrangements of α -Hydroxy Ketals and Derivatives of α -Hydroxy Ketals.

Page 4231. The section of this paper dealing with the solvolysis of 23 concluded that process involved a stepwise assisted ionization leading ultimately to the same product seen by Conia in the pyrolysis of 1. This paper may have inadvertently implied that similar processes

$$\begin{array}{c|c} OCH_3 & OCH_3 \\ \hline OCH_3 & OCH_3 \\ \hline OTs & \\ \end{array}$$

have been described by Conia as concerted. We do not mean to imply this and, indeed, Conia has elegantly shown that these processes are probably not concerted [J. Salaun and J. M. Conia, Tetrahedron Lett., 4023 (1971)]. We simply meant to show the similarity in products from solvolysis and pyrolysis and meant to imply nothing concerning the pyrolysis mechanism. Additionally our studies on the rearrangement of 2,2-dimethoxycyclobutanol were meant to verify the involvement of the aldehyde intermediate 16 suggested by Conia in the formation of 6 by the deuterium labeling study. Based on the result of this experiment, we feel that Conia's later suggested nonconcerted process is correct and so state in this paper.

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P. Narayanan, F. Ramirez, T. McCaffrey, Y. F. Chaw, and J. F. Marecek: Crystal and Molecular Structure of Pentaaquohexa-(diphenylphosphato)trimagnesium(II), $[(C_6H_5O)_2P(O)O]_6Mg_3$ -(H2O)5, a Hydrated Magnesium Phosphodiester Salt with Penta- and Hexacoordinate Metal Ions.

Page 24. Abstract, line 4. For " $Z = 2\lambda$ " read "Z = 2". Page 28. Line 11. For "ideal trigonal pyramid" read "ideal tetragonal pyramid"

Page 30. Reference 20 should read: (20) (a) M. A. Viswamitra, M. V. Hosur, Z. Shakked, and O. Kennard, Nature (London), 262, 234 (1976); (b) O. Kennard, N. W. Isaacs, W. D. S. Mutherwell, J. C. Coppola, D. L. Wampler, A. C. Larson, and D. G. Watson, Proc. R. Soc. London, Ser. A, 325, 401 (1971).

Page 30. Reference 22 should read: (22) M. A. Viswamitra, T. P. Seshadri, M. L. Post, and O. Kennard, Nature (London), 258, 497

Tsuruji Goka, Maruo Shizuka,* and Kohji Matsui: Photolyses of 2-Azido-4-methoxy-6-(1-naphthyl)-1,3,5-triazines: Reactions of 1,3,5-Triazenylnitrenes with Solvents.

Page 1364. Table V contains two errors in the equation: CH₃COCH₃ should be CH₃CN, and the adduct is shown with one too many methyl groups on the lower C2N3 ring.

Louis D. Quin* and Shin Ok Lee: Stereochemical Consequences of C-Methylation of 1-Methylphosphorinane and Its Sulfide and Oxide: A Carbon-13 and Phosphorus-31 Nuclear Magnetic Resonance Study.

Page 1429. Reference numbers 14 through 19 should be changed to 15 through 20 to match usage in the text. New reference 14, which was inadvertently omitted in a revision, should be: M. D. Gordon and L. D. Quin, J. Org. Chem., 41, 1690 (1976).

Arthur G. Anderson, Jr.,* David M. Forkey, and Larry D. Grina: Reactions of 2-Methyl-2H-cyclopenta[d]pyridazines with Nitration Reagents, Mercuric Acetate, and Tetracyanoethene.
Page 1602. Line 9 of text now reads, "... that direct nitration would

occur with difficulty", and should read, "... that direct nitration would occur without difficulty".

David K. Minster, Ulrich Jordis, David L. Evans, and Sidney M. Hecht*: Thiazoles from Cysteinyl Peptides.

Page 1625. Column 1, lines 10-14. The sentence beginning on the tenth line of text should read "Although preparations of nickel peroxide contain fewer oxidizing equivalents per gram of catalyst than does MnO₂, as judged by release of I₂ from potassium iodide solution, we reasoned that the greater oxidizing power (or possibly instability) of Ni(IV) as compared with Mn(IV) should make NiO2 the more effective oxidant.

Gerald J. Boudreaux, Ernest I. Becker,* and Byron H. Arison: Intramolecular Aldol Condensation of 2,2'-Dimethylbenzil.

Page 1828. Scheme I. In the two structures that appear in the fourth row of the scheme, the two tolyl groups should be ortho instead of meta, as shown below:

$$\begin{bmatrix} 0 & 0 \\ CH_3 & CH_3 \end{bmatrix} + H_2O \Rightarrow CH_3 + HO^{-1}$$
2-bydroxy-2-(0-tolyl)-1-

2-oxido-2-(o-tolyl)-1indanone

2-hydroxy-2-(o-tolyl)-1indanone

Robert O. Hutchins* and Nicholas R. Natale: Sodium Borohydride in Acetic Acid. A Convenient System for the Deoxygenation of Carbonyl Tosylhydrazones.

Page 2299. Column 2, in eq 1 the product olefin, incorrectly shown as 1-pentene, should be 2-pentene.

Page 2299. Table I, footnote b should read "... The tosylhydrazone prepared in situ in acetic acid from the ketone and tosylhydrazine

Page 2300. Column 1, eq 2, the upper middle structure, incorrectly shown as 3-methyl-1-butene, should be the carbonium ion, $CH_2 = CH - + CHCH_3$.